



SIZE-DEPENDENT EMERGENCE OF CRYSTALINITY IN MOLECULAR CLUSTERS

Ana Proykova

Head, High Performance Computing Laboratory,
Sofia University & Sofia Tech Park

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with International Participation
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Sofia University St. Kliment Ohridski 1888



Laboratory complex, Sofia Tech Park 2016



Layout

Why clusters?

Production and experimental study

Computational methods and computers
(clusters)

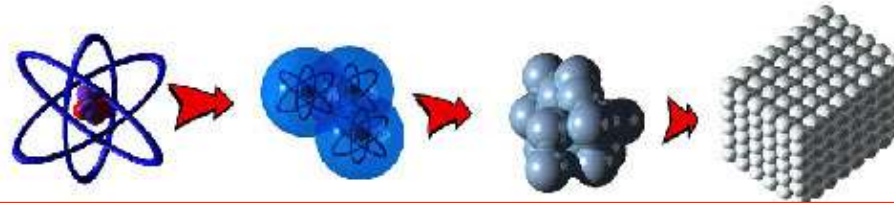
Phase transitions – disorder-order (cooling)

Phase coexistence (only clusters, not in bulk)

Topography of the potential energy surfaces

What can we learn when studying clusters

- 1960-1975: from atom to bulk



- 1975 – at present
Cluster physics to nanoscience

Clusters of atoms – stability and symmetry

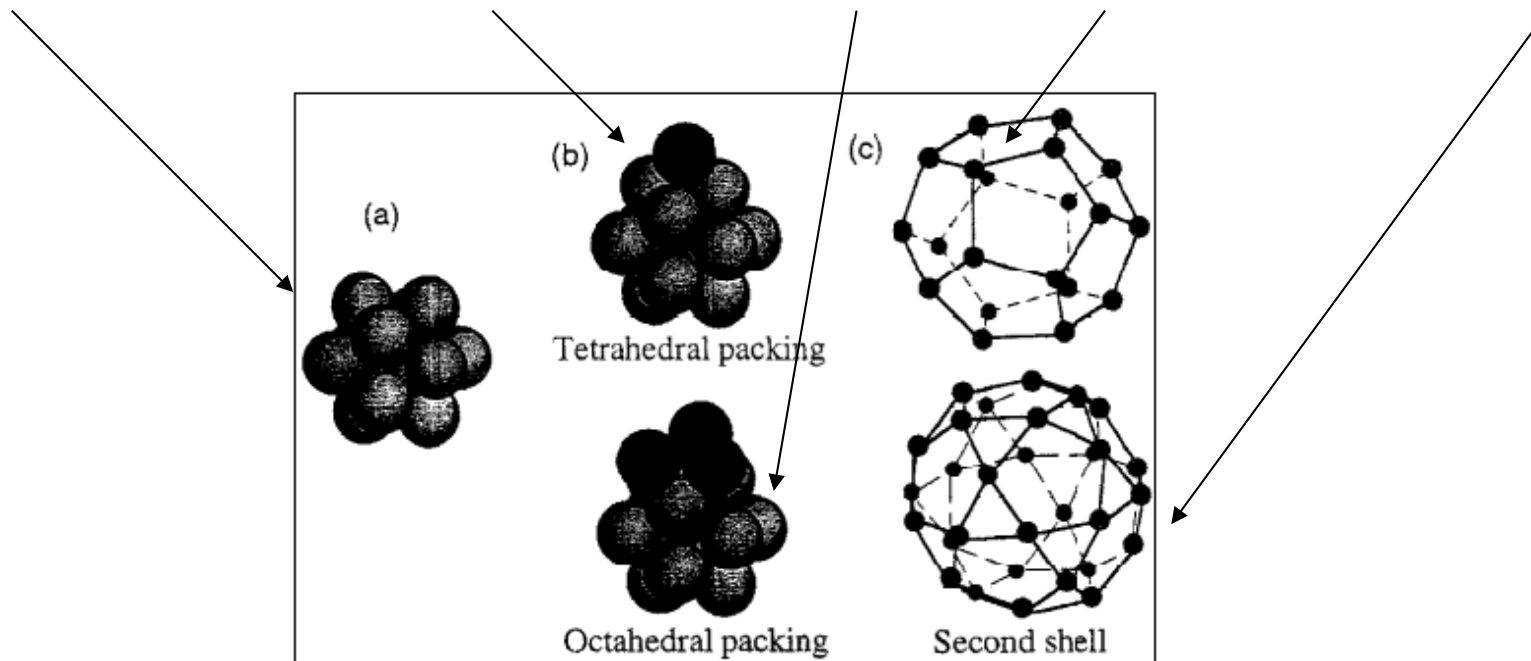
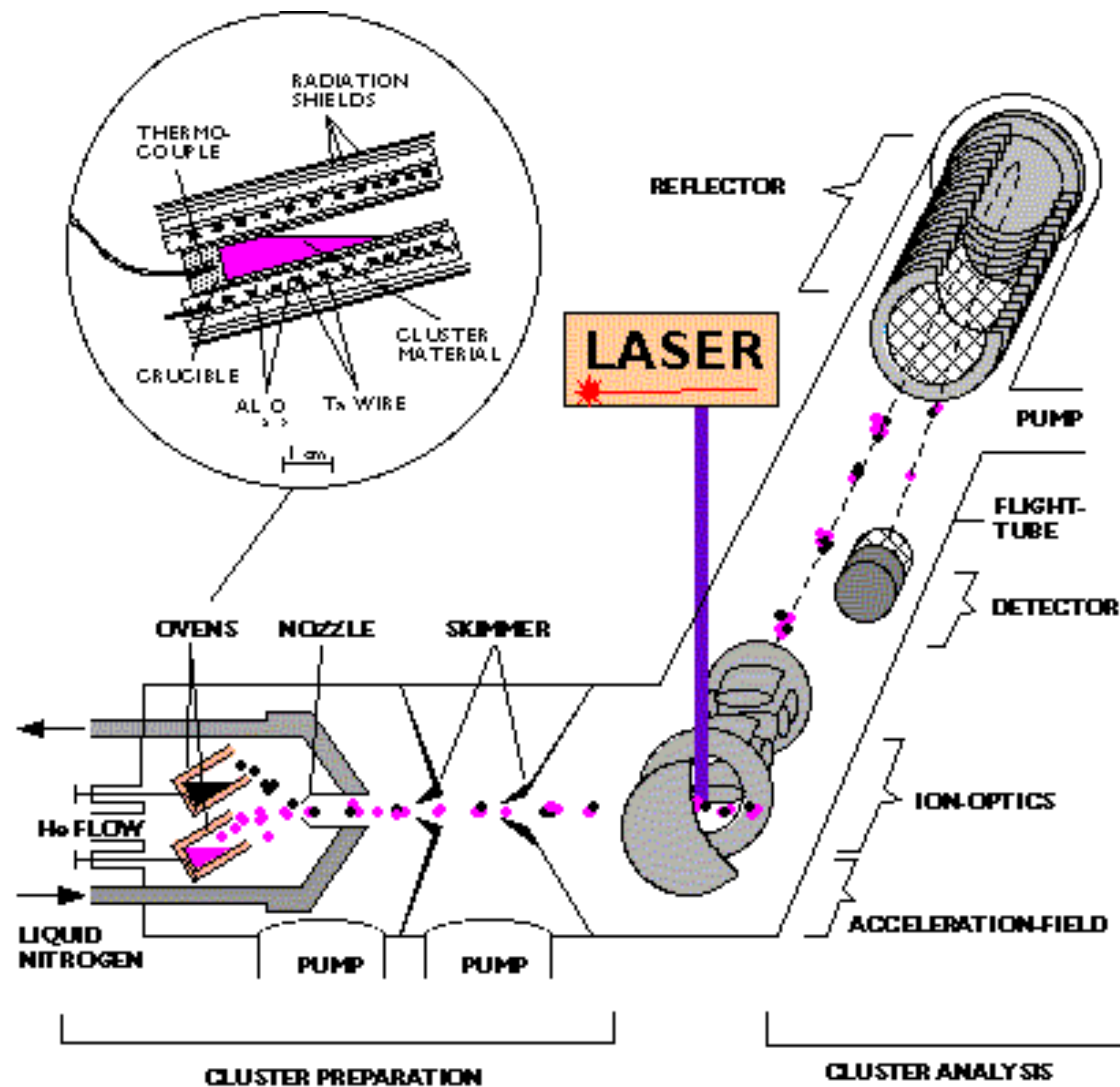


Fig. 3. (a); A fundamental atom cluster, (b); tetrahedral and octahedral packings of atoms on (a), and (c); second atom shells formed by tetrahedral and octahedral packings.

Molecular clusters

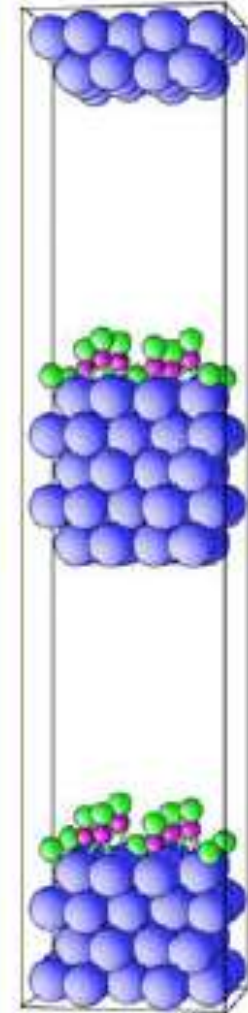
- Clusters consisting of N octahedral molecules AF₆ exhibit several structural phase transformations under the temperature of solidification – this is shown with Monte Carlo simulations, Molecular Dynamics calculations, and quantum computations (density-functional theory) and experiments.

Laser production of clusters (nozzle-jet) & analysis with electron diffraction



Methods

- I** {
- **Density functional theory**
 - *ab initio* pseudopotentials
(the fhi98md - code ---
www.fhi-berlin.mpg.de/th/th.html)
 - FP-LAPW
(the WIEN - code by
P. Blaha, K. Schwarz, et al.;
M. Petersen et al., CPC 126 (2000))
- II** {
- *ab initio* Molecular Dynamics
 - *ab initio* Quantum Dynamics
 - *ab initio* Lattice Gas Hamiltonian
 - *ab initio* kinetic Monte Carlo





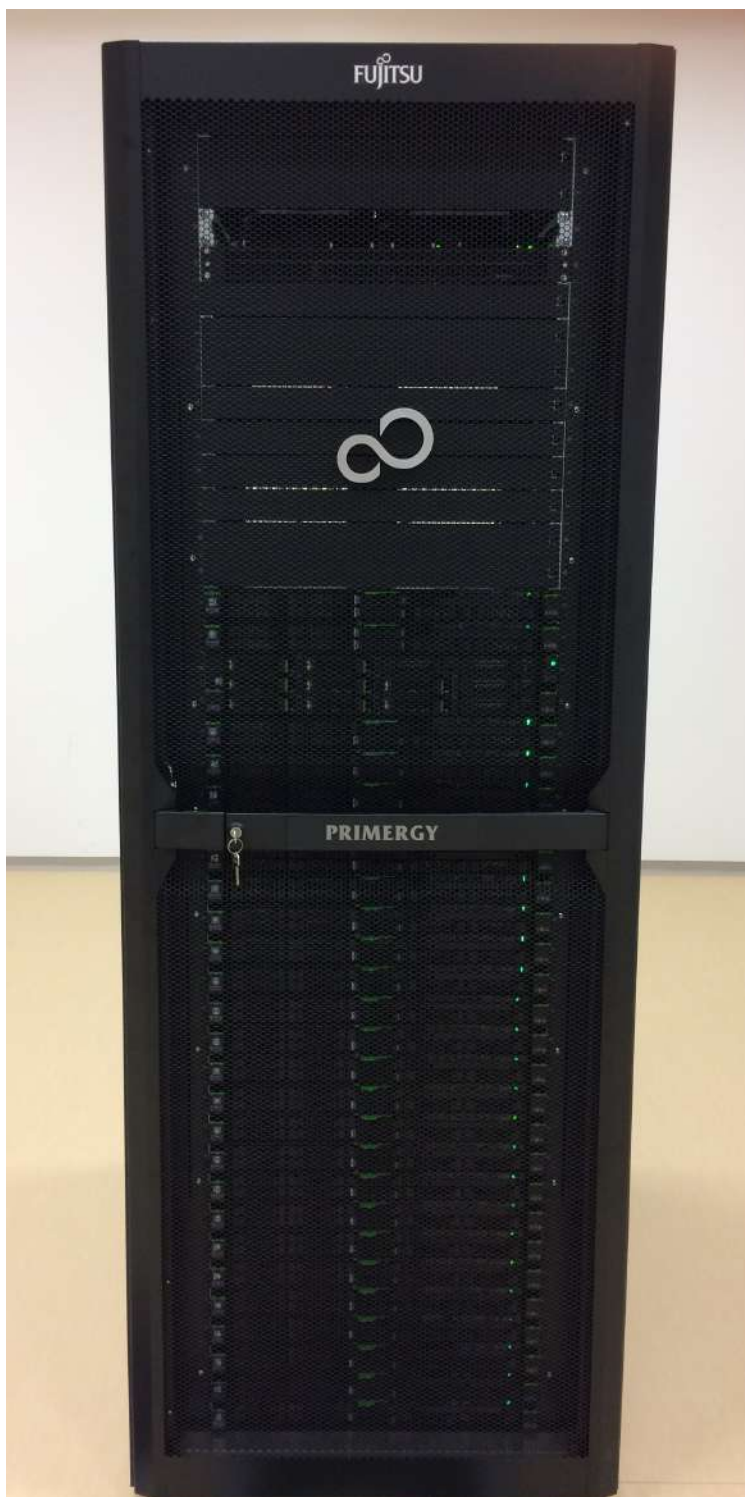
PhysOn Cluster

<http://physon.phys.uni-sofia.bg>



PhysOn – initial stage



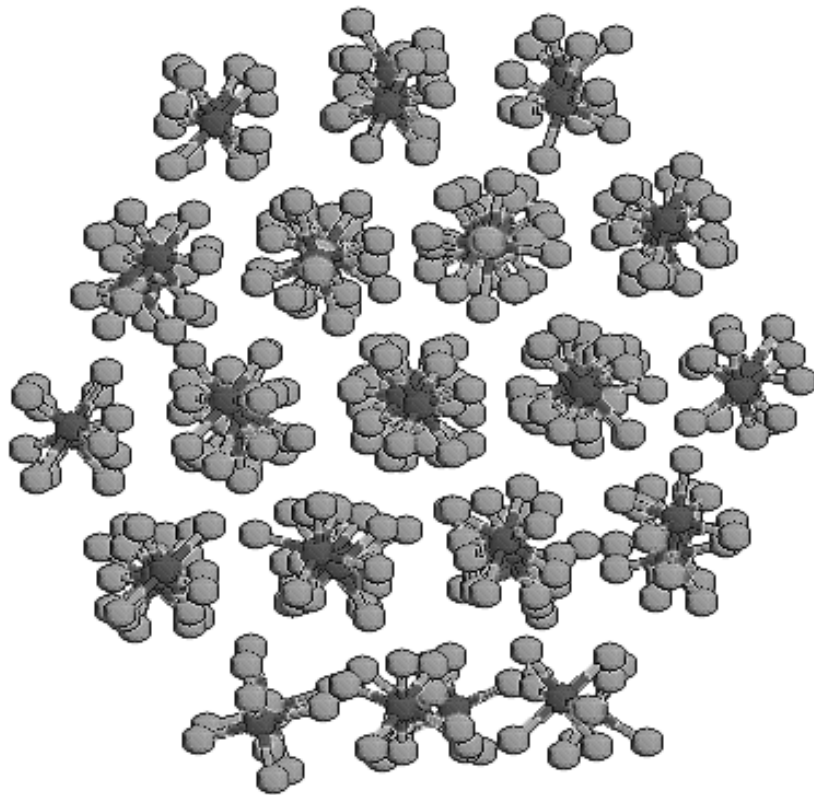


Nestum Cluster

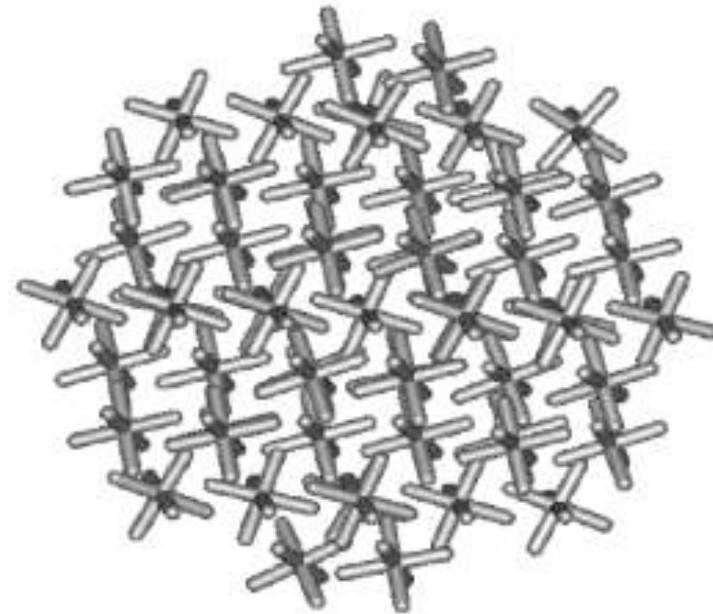
- A homogeneous cluster based on Intel Xeon Haswell (22 nm processes) architecture
- 24 nodes with two 16-core processors Intel Xeon E5-2698 v3 @ 2.3 GHz
- Infiniband with 56 Gbps
- SAN (storage area network) subsystem with 10 TB; Raid 50
- Dynamical memory 3TB

Cluster of 59 octahedral molecules – molecular dynamics simulations at different temperatures

at T = 100 K (bcc)



at T = 20 K (bcc)



:

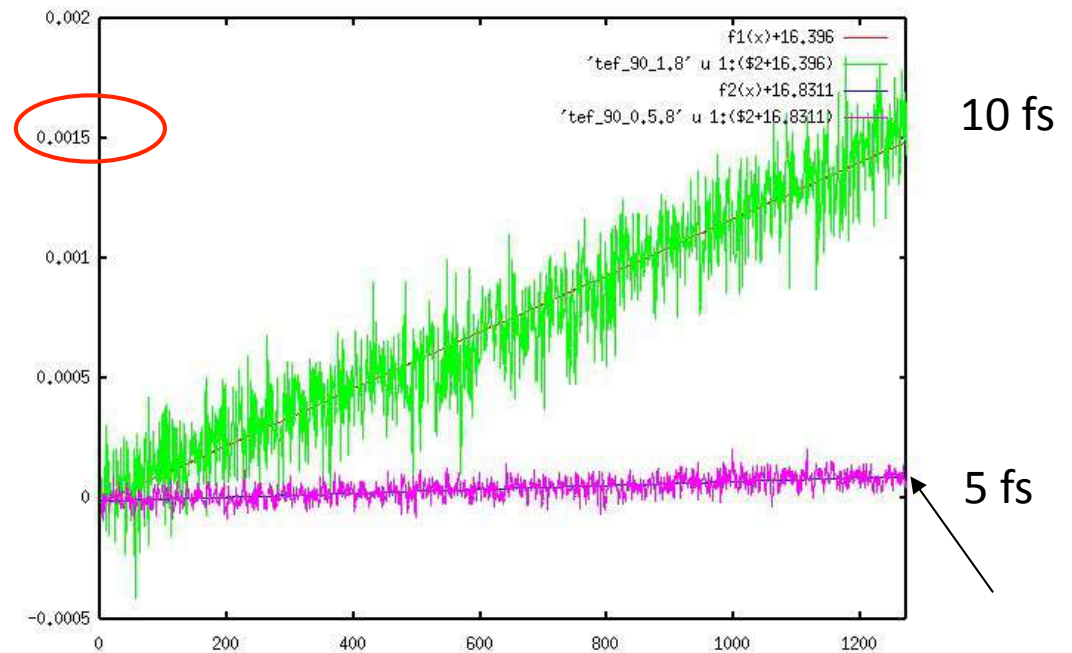
Molecular dynamics – equivalent to solving the Hamilton equations
 $6(N-1)$, $7 < N < 300$

$$dr_i/dt = \partial E/\partial p_i$$

$$dp_i/dt = -\partial E/\partial r_i$$

a) $E = U(r) + E_{kin}(p) = \text{const}$

- Microcanonical ensemble



Temperature – kinetic energy

- Linear and angular velocity – re-scaling with s ;
 g – degrees of freedom

s

$$s^2 \sum_i 0.5 m_i [v_i(t + \Delta t / 2)]^2 = (g / 2) k_B \langle T_{vib} \rangle$$

$$s^2 \sum_i 0.5 m_i [\omega_i(t + \Delta t / 2)]^2 = (g / 2) k_B \langle T_{rot} \rangle$$

Transfer between two coordinate systems

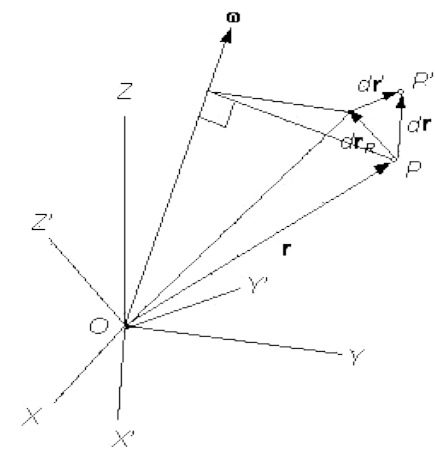
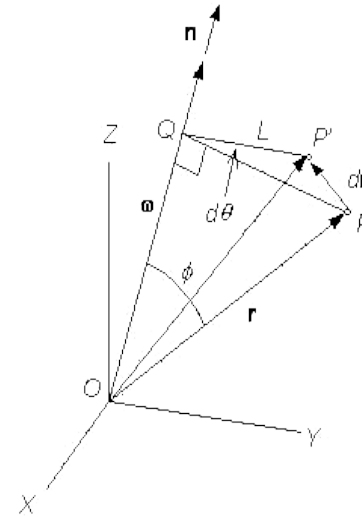
b- molecular s - cluster

$$\tau_i = \sum_a (\mathbf{r}_{ia} - \mathbf{r}_i) \times \mathbf{f}_{ia} = \sum_a \mathbf{d}_{ia} \times \mathbf{f}_{ia}.$$

$$\tau^b = A \cdot \tau^s$$

$$\omega^s = A^{-1} \cdot \omega^b = A^T \cdot \omega^b$$

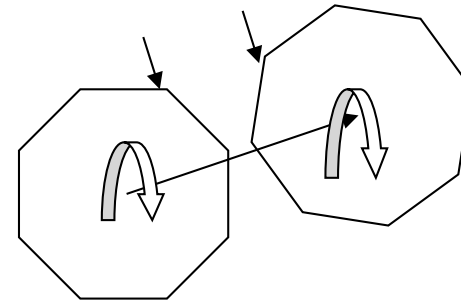
$$A = \begin{pmatrix} q_0^2 + q_1^2 - q_2^2 - q_3^2 & 2(q_1q_2 + q_0q_3) & 2(q_1q_3 - q_0q_2) \\ 2(q_1q_2 - q_0q_3) & q_0^2 - q_1^2 + q_2^2 - q_3^2 & 2(q_2q_3 + q_0q_1) \\ 2(q_1q_3 + q_0q_2) & a(q_2q_3 - q_0q_1) & q_0^2 - q_1^2 - q_2^2 + q_3^2 \end{pmatrix}.$$



Total angular momentum \mathbf{J} of a plastic cluster made of octahedral molecules that rotate and vibrate

- $$\mathbf{J} = \sum_i m \mathbf{r}_i \times \mathbf{v}_i + \sum_i I \boldsymbol{\omega}_i = L_{clust} + \sum_i I \boldsymbol{\omega}_i$$
- $\boldsymbol{\omega}_i$ Angular velocity of the i -th molecule in the molecular system
- The lab (space) coordinate system is centered in the cluster center of mass with a fixed orientation

Let us consider two points of the molecules which come nearer as a result of interaction; the molecules rotate with $\boldsymbol{\omega}_i$ and $\boldsymbol{\omega}_j$, respectively



Total angular momentum (2)

- The vector of the relative velocity is

$$v_{ij}^{imp} = v_i^{imp} - v_j^{imp} = (v_i - v_j) - 1/2(\omega_i + \omega_j) \times r_{ij}$$

- Each 'bumpy' molecule reverses the sign of velocity as a result of collision:

$$\delta v_{ij}^{imp} = v_{ij}^{imp} (af\ te\ \grave{e}) - v_{ij}^{imp} (bef\ or\ \grave{e}) = -2v_{ij}^{imp} (bef\ or\ \grave{e})$$

- Impulse changes to:

$$\delta p_i = \frac{1}{2} m \left(\delta v_{ij}^{imp\parallel} + \frac{k}{1+k} \delta v_{ij}^{imp\perp} \right) \quad k = 4I / 2mR$$

$$v_{ij}^{imp} = v_{ij}^{imp\perp} + v_{ij}^{imp\parallel}$$

Angular momentum (3)

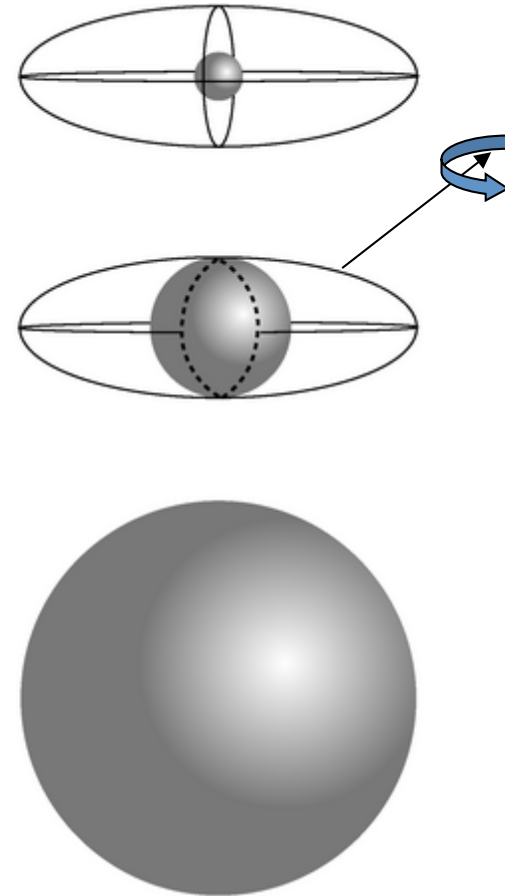
- The total energy

$$E_L^{\text{"rb"}}(t) = E_L^{\text{"rb"}}(\{r_i(t)\})$$

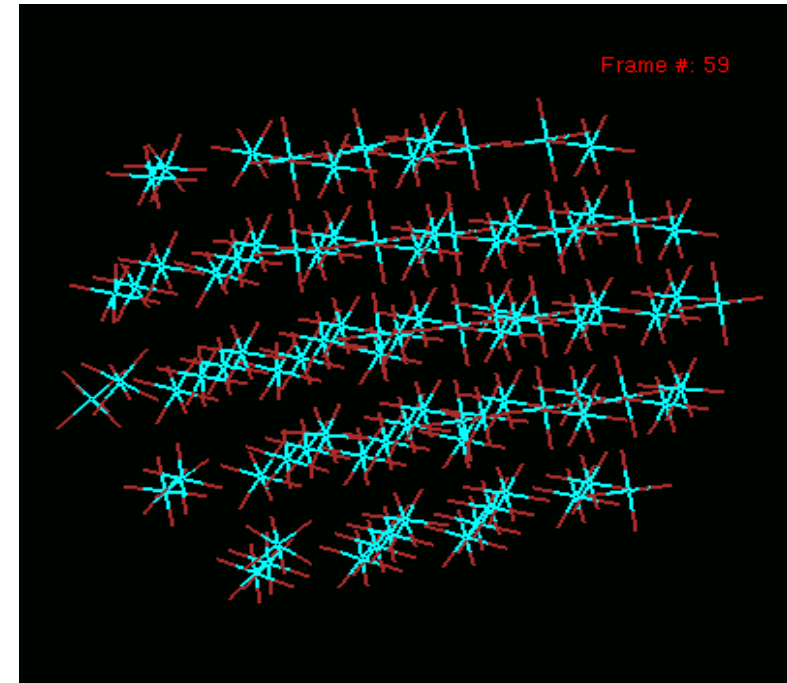
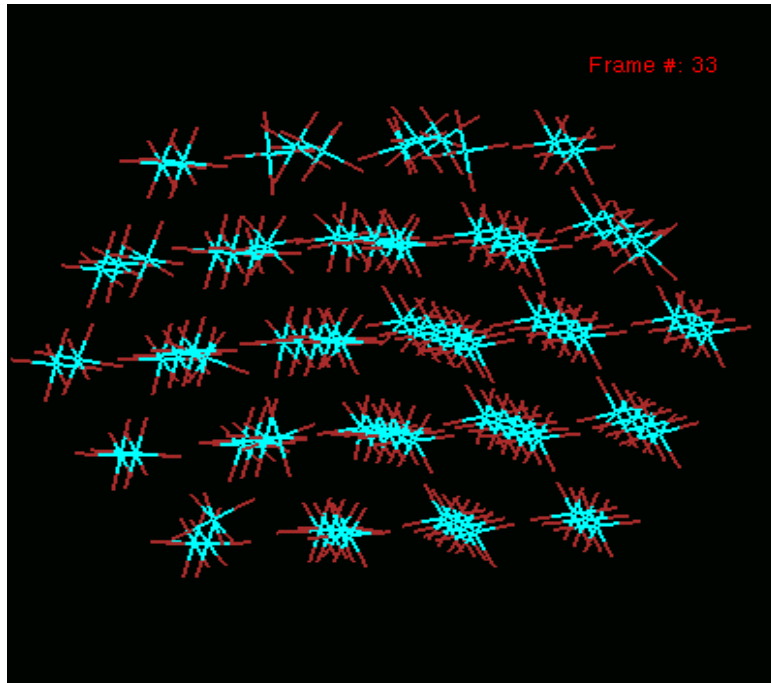
$$\sum_i \frac{p_i^2}{2m_i} = E_L^{\text{"rb"}} + \sum_i \frac{(\Delta p_i)^2}{2m_i}$$

- Rotation-vibration potential

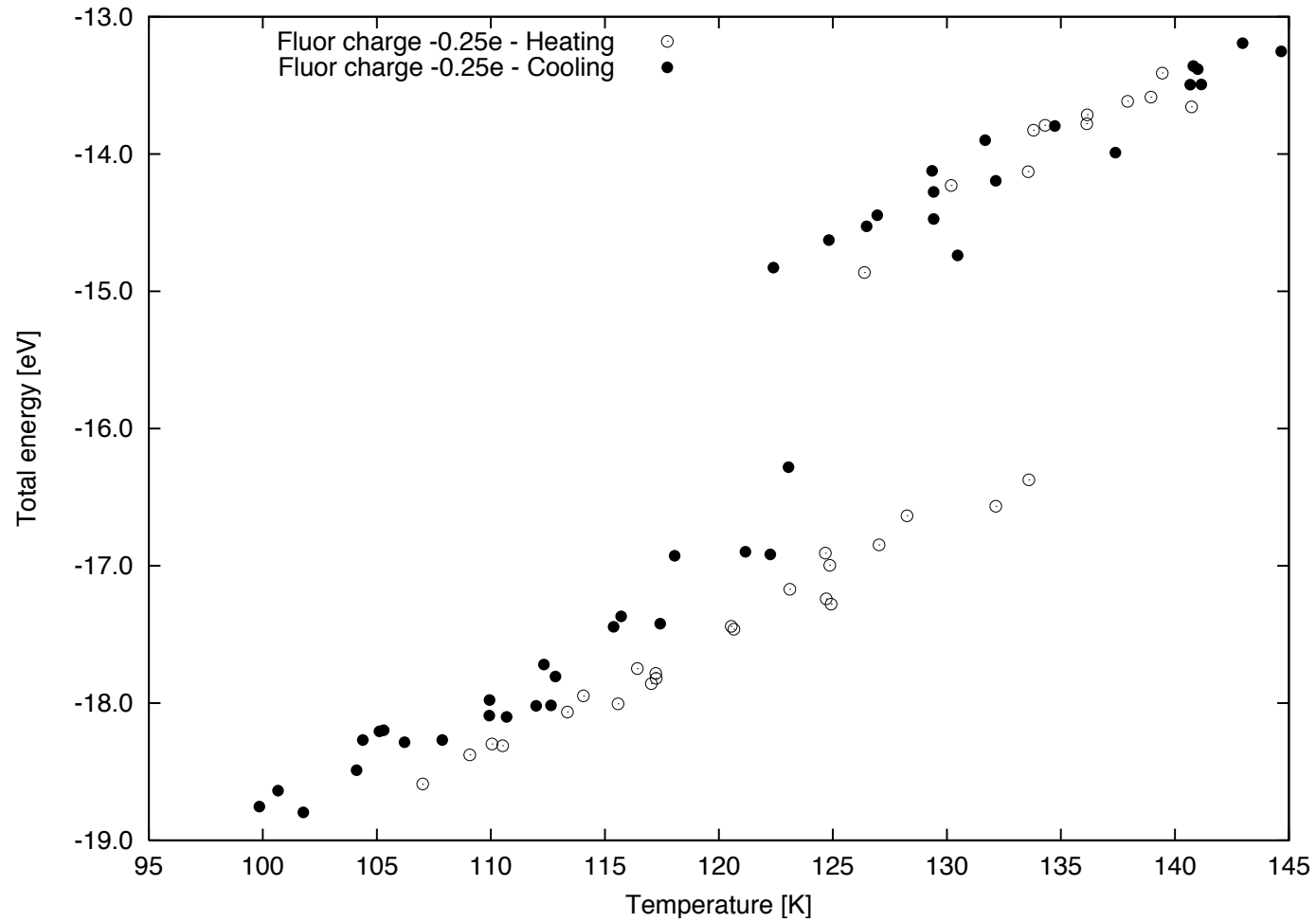
$$V_L(\{r_i\}) = E_L^{\text{"rb"}}(\{r_i(t)\}) + V(\{r_i\})$$



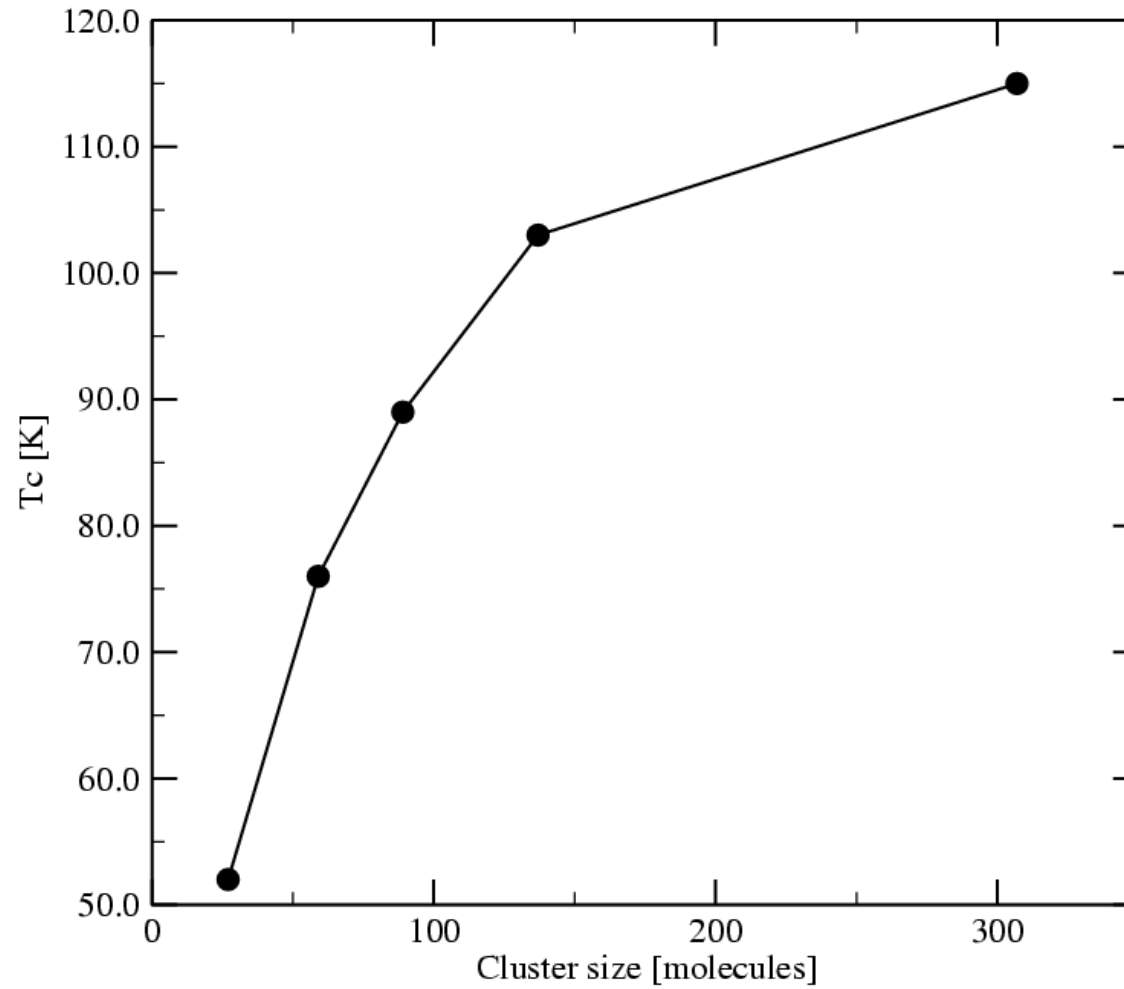
Coexistence of different crystal phases



Phase transformations – latent heat



Size-dependent critical temperature



Potential energy surface - resonant inelastic X-ray scattering (RIXS)

<https://www.tandfonline.com/doi/full/10.1080/08940886.2017.1289799#>

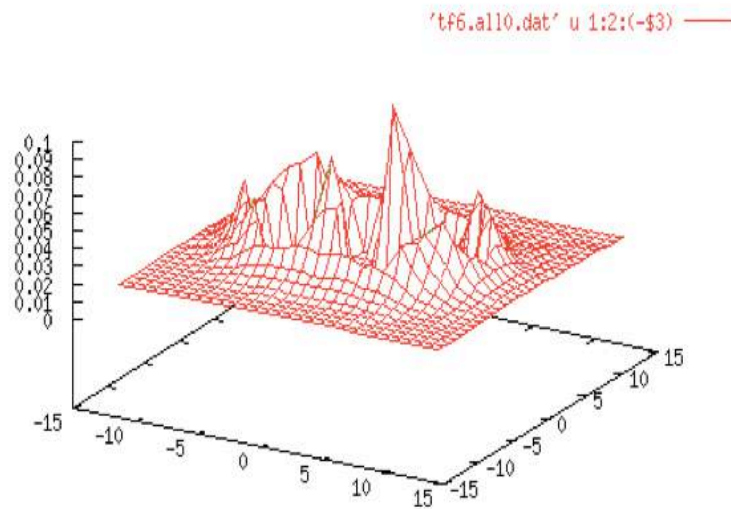
Potential energy surfaces are a central tool to rationalize, in a multidimensional system, how distortions along defined degrees of freedom link conformation and structural modifications to energy. The local and global minima define stable conformations and the ground state.

Optically excited or spin-excited states live on different potential energy surfaces.

Topography - computed

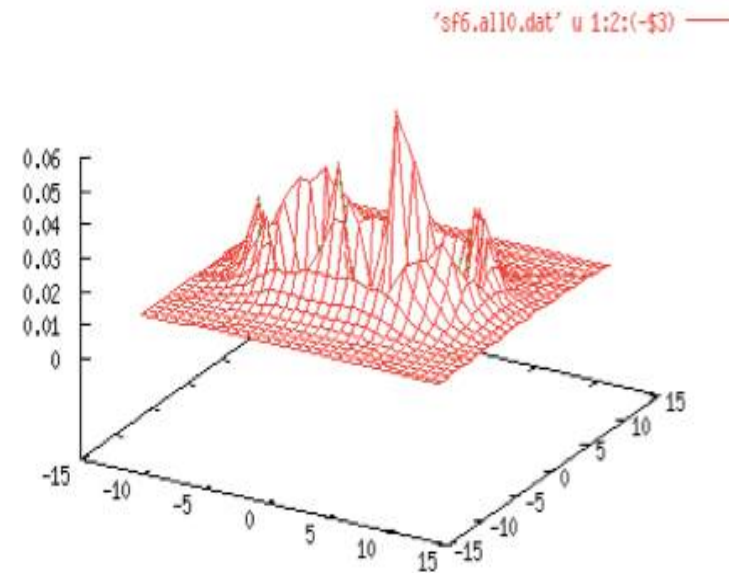
TeF6

- Max 0.1

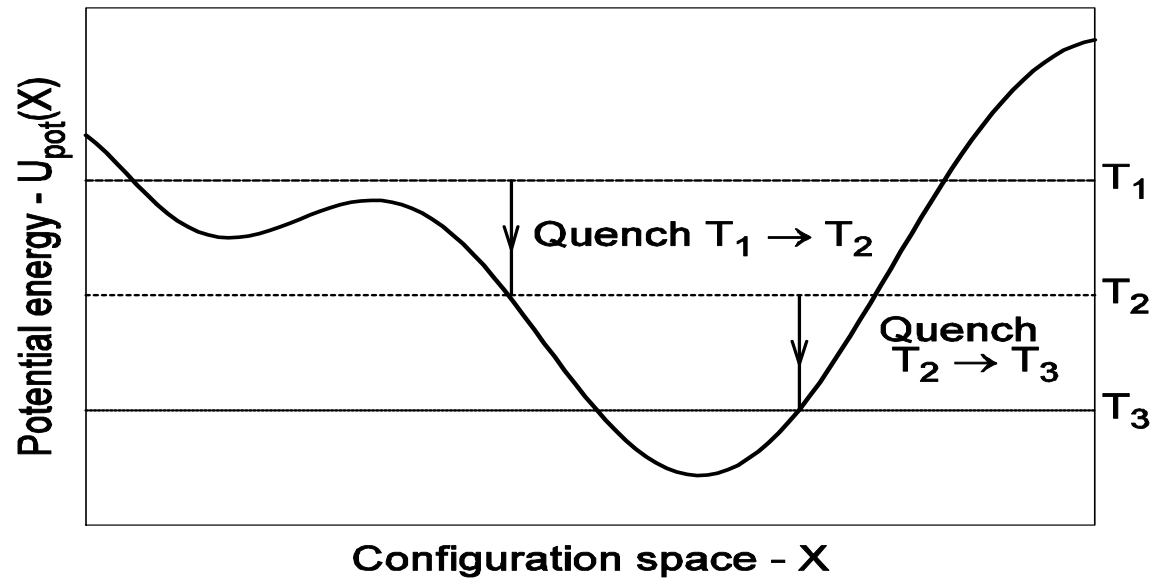


SF6

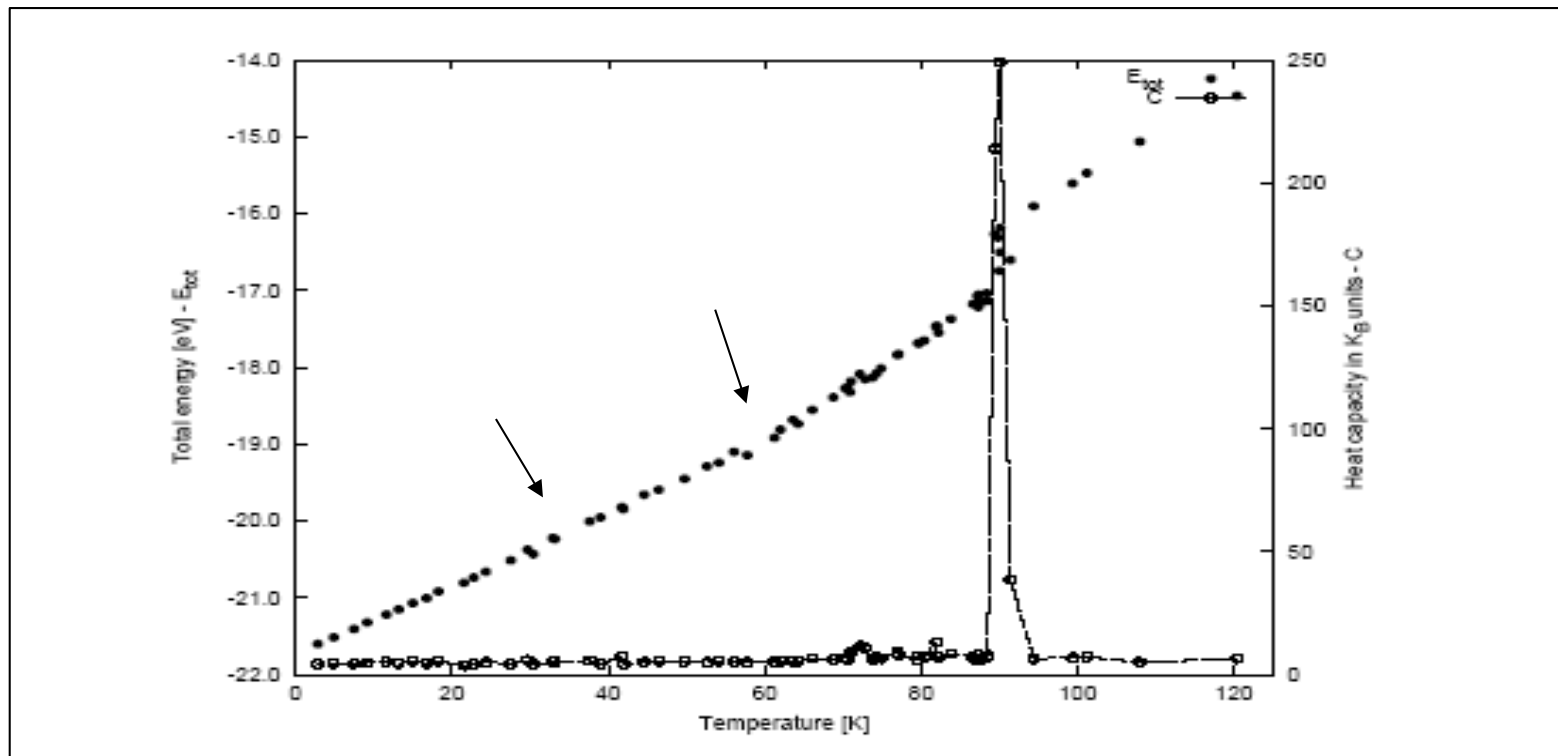
- Max 0.06



Local minima search - quenching



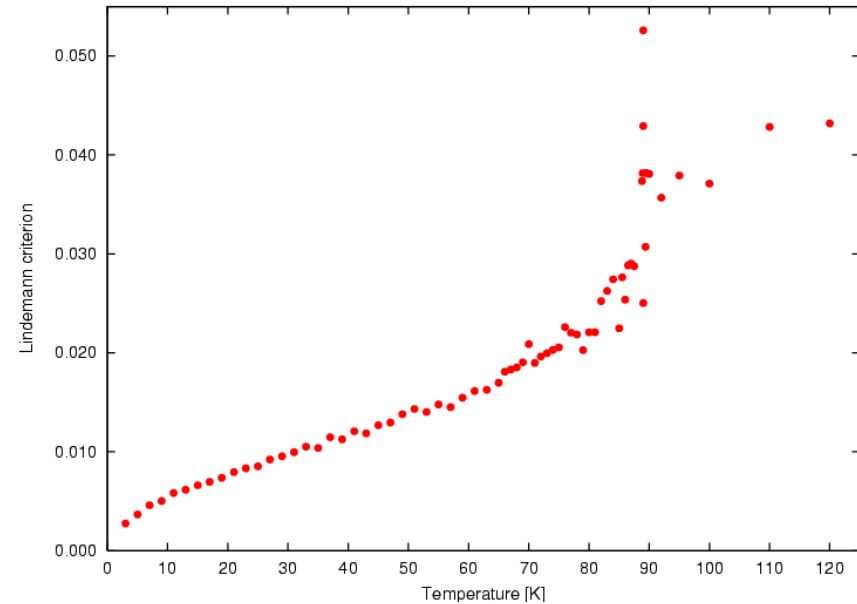
$$\frac{\langle (\delta K)^2 \rangle}{\langle K \rangle} = \frac{N}{N-1} k_B T \left(1 - \frac{3k_B}{2C} \right)$$



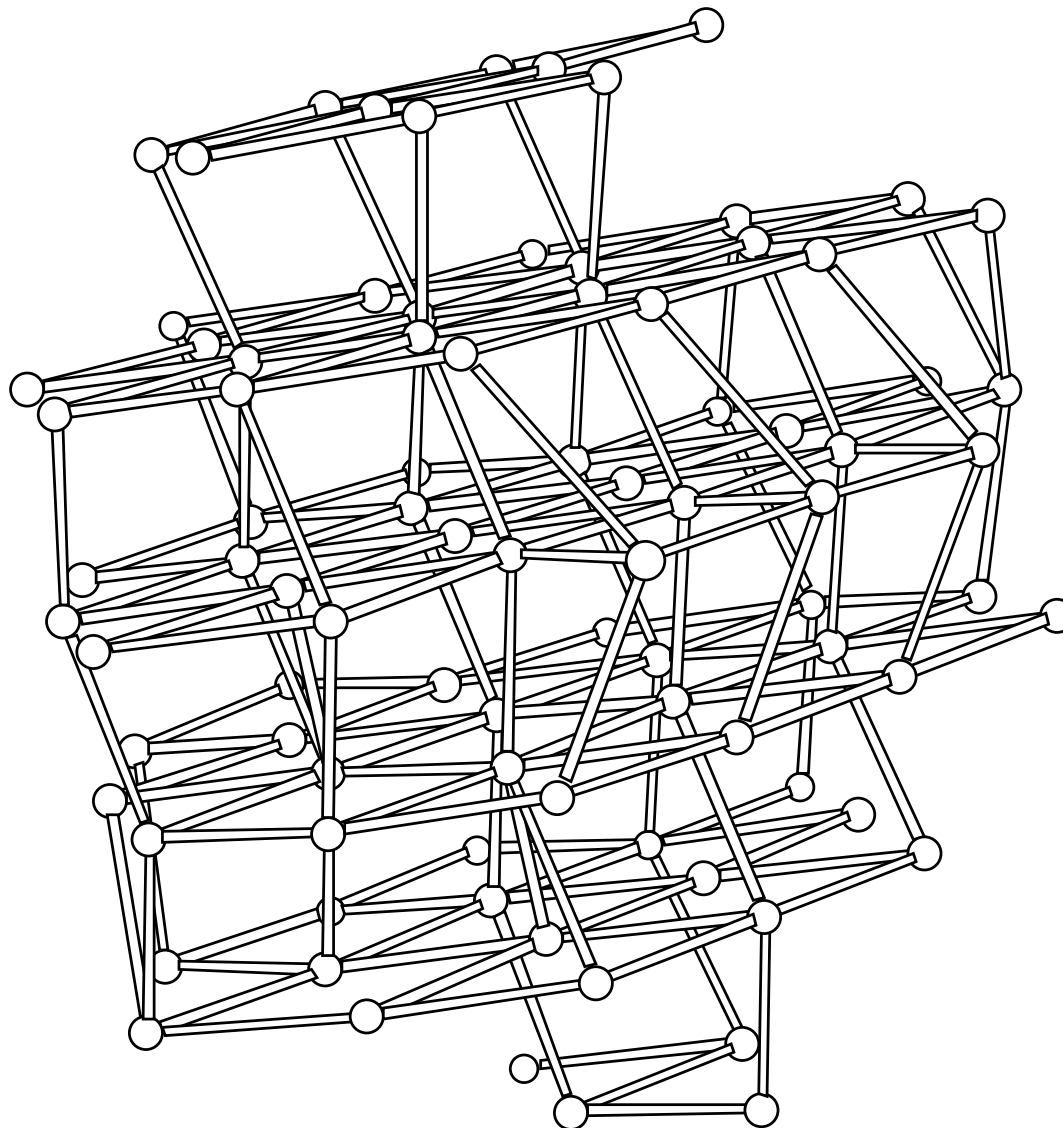
Liquid-like or solid-like

- Lindeman criterion

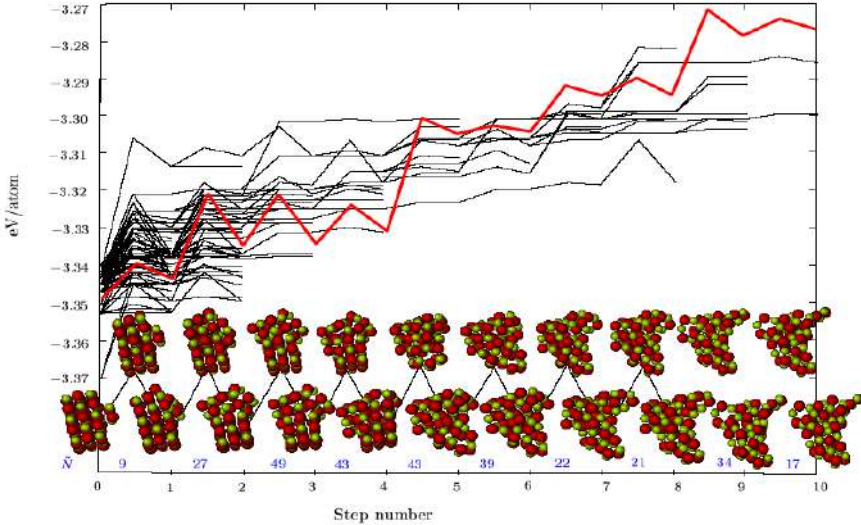
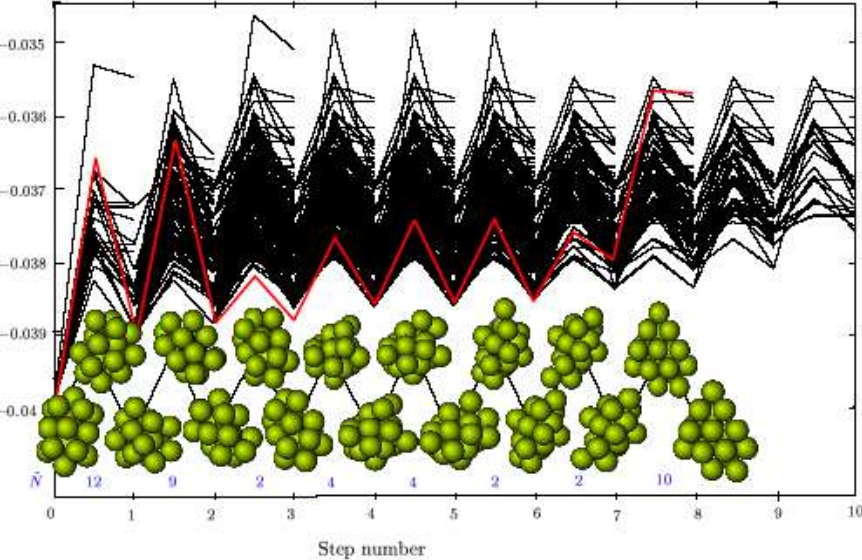
$$\delta_{lin} = \frac{2}{N(N-1)} \sum_{i,j(>i)=1}^N \frac{\sqrt{\langle r_{ij}^2 \rangle - \langle r_{ij} \rangle^2}}{\langle r_{ij} \rangle}$$



Centers of Te (TeF₆) – T = 20 K

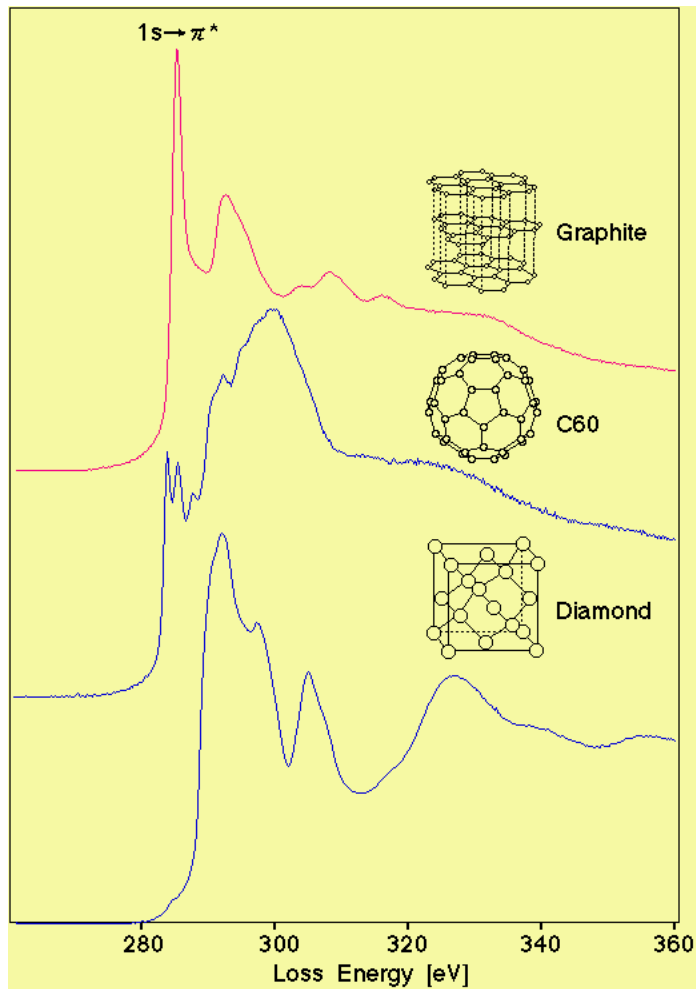


The topography of the potential energy surface determine the lowest energy state: glass or crystal



Example

EELS = electron energy loss spectroscopy:



- Low energy electrons: scattering with carbon objects

Acknowledgement

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NESTUM cluster

- Calculations of unusual magnetic and electronic properties of carbon materials
- Studying peculiarities of crystal growth
- Climate change predictions
- Big Data for Smart Society
- Big Data Analytics – Financial market



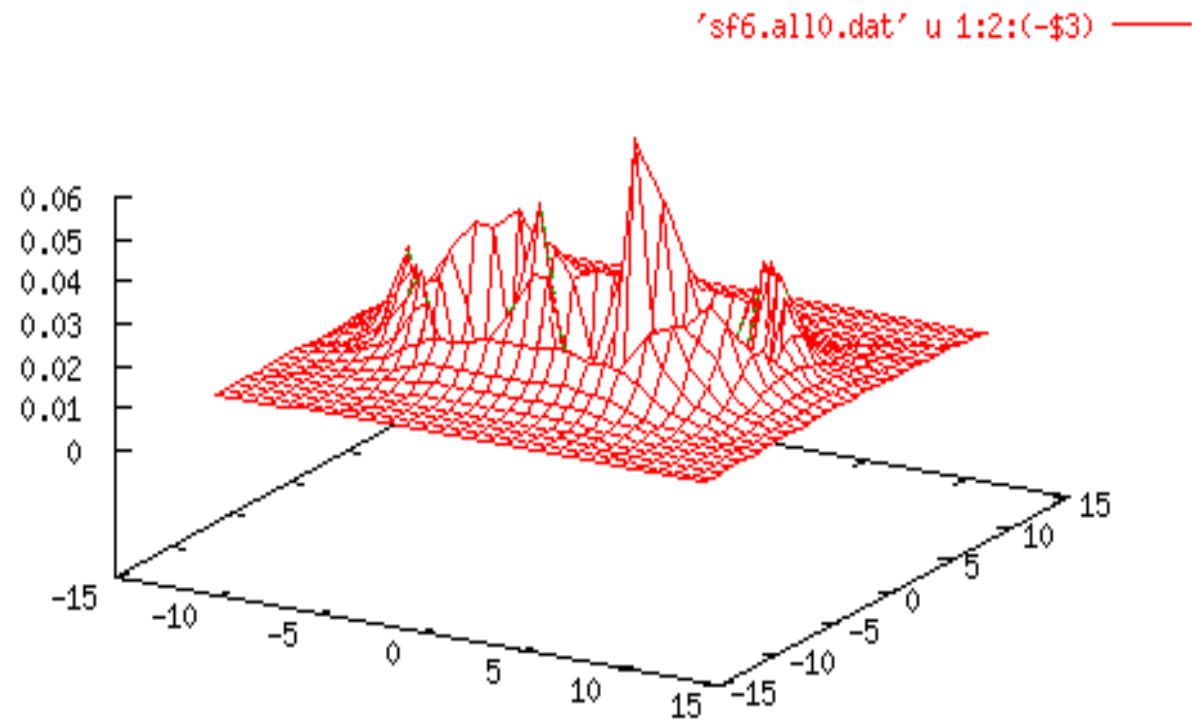
Thank you for listening!

Благодаря за вниманието!

anap@phys.uni-sofia.bg

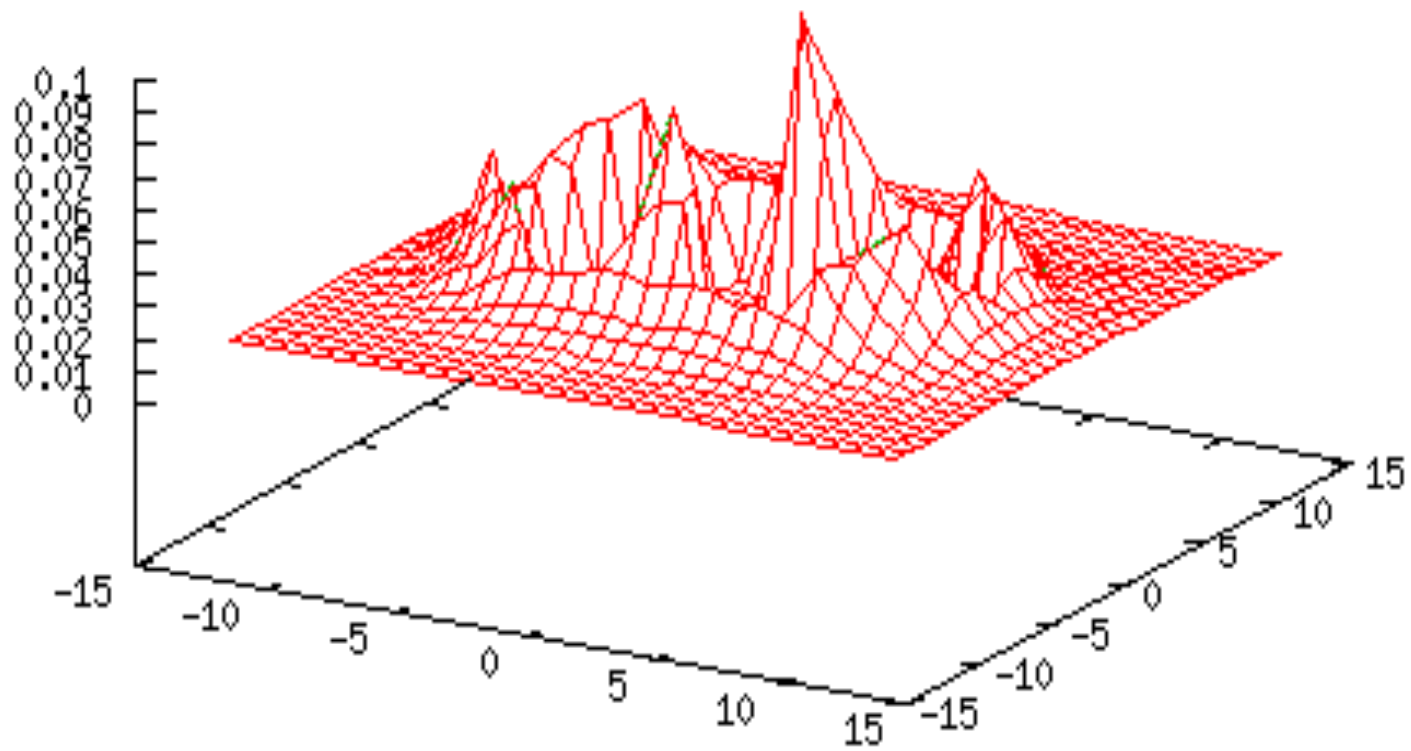
ana.proykova@fulbrightmail.org

SF6 surface



TeF6 surface

'tf6.all0.dat' u 1:2:(-\$3) —



Methane cluster (computed)

